

Problem 1

The classical amplitude is given by $x_0 = A$, satisfying

$$E = \frac{1}{2} m_e \omega^2 A^2$$

In the ground state, $E = E_0 = \frac{\hbar \omega}{2}$, so, with $A =$

$$\hbar \omega =$$

Harmonic oscillator only absorbs photons of frequency ω , wavelength

$$\lambda, \text{ with } \frac{hc}{\lambda} = \hbar \omega \Rightarrow$$

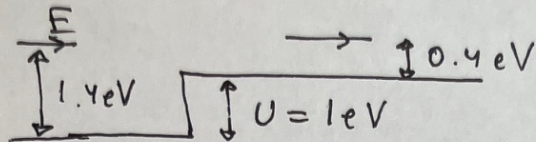
(b) Before absorbing radiation, electron in ground state with

$$\Psi_0(x) = C_0 e^{-x^2/2A^2} \Rightarrow \text{most likely to be found at}$$

After absorbing a photon, electron in state

$$\Psi_1(x) = C_1 x e^{-x^2/2A^2} \quad \text{Probability is}$$

Problem 2



incident energy is $E = 1.4 \text{ eV}$

Reflection coef $R = \frac{(A_1 - A_2)^2}{(A_1 + A_2)^2}$

$$A_1 = \sqrt{\frac{2m_e E}{\hbar^2}} =$$

$$A_2 = \sqrt{\frac{2m_e (E - U)}{\hbar^2}} =$$

$$R =$$

$$\text{transmission coef} = T = 1 - R =$$

If 1000 electrons were transmitted, the number of incident electrons was N such that $N \cdot T = 1000 \Rightarrow$

Problem 3

According to the Rutherford formula, # of α particles scattered at angle θ is proportional to $\frac{1}{\sin^4 \theta/2}$

$$\Rightarrow \frac{N(\theta=90^\circ)}{N(\theta=180^\circ)} = \frac{\sin^4(45^\circ)}{\sin^4(90^\circ)}$$

So according to the problem, this is what is observed for kinetic energy of α particle $K_\alpha = 4 \text{ MeV}$.

The distance of closest approach to the center of the nucleus for impact parameter $b=0$ is r , satisfying

$$K_\alpha = \frac{2ze^2}{4\pi\epsilon_0 r} = 1$$

$$\Rightarrow r =$$

For the kinetic energies given in the problem

K_α	r	Is Rutherford law satisfied? i.e. if α particles don't penetrate the nucleus
3.5 MeV		yes
4 MeV		
4.5 MeV		no
5 MeV		

(a) From the fact that Rutherford law is satisfied for $K_\alpha = 4 \text{ MeV}$ and not satisfied for $K_\alpha = 5 \text{ MeV}$ we deduce that the radius of the nucleus of Ag, r_{Ag} , is $r_{\text{Ag}} =$

(b) For $K_\alpha = 3.5 \text{ MeV}$, the α particles stay outside nucleus $\Rightarrow N(90^\circ) =$

For $K_\alpha = 4.5 \text{ MeV}$

Problem 4

$$E = -13.6 \text{ eV} = -E_0, \quad r = 0.2116 \text{ nm} = 4a_0$$

We know that:

$$E_n = -\frac{E_0 Z^2}{n^2} = -E_0 \Rightarrow$$

$$\text{and that } r_n = \frac{a_0}{Z} n^2 = 4a_0 =$$

=

(a) angular momentum is $L_n = n\hbar$

(b) The shortest wavelength is for the transition $n=4$

$$\frac{hc}{\lambda} = E$$

Problem 5

$$F = -kr \Rightarrow$$

$$kr = \frac{m_e v^2}{r} \Rightarrow \boxed{kr^2 = m_e v^2}$$

$$\Rightarrow \boxed{K = \text{[redacted]}}$$

(b) Assuming $L = n\hbar$ as in the Bohr atom,

$$L = m_e v r = n\hbar \Rightarrow L^2 = m_e^2 v^2 r^2 = n^2 \hbar^2 \Rightarrow$$

[redacted]

Smallest orbit is for $n=1$

$$r_1 =$$

$$\text{For } k = \frac{1 \text{ eV}}{\text{nm}^2}, \quad \boxed{r_1 = \text{[redacted]}}$$