

PHYSICS 140B : STATISTICAL PHYSICS
HW ASSIGNMENT #1

(1) Consider a two-dimensional gas of fermions which obey the dispersion relation

$$\varepsilon(\mathbf{k}) = \varepsilon_0 \left((k_x^2 + k_y^2) a^2 + \frac{1}{2} (k_x^4 + k_y^4) a^4 \right) .$$

Sketch, on the same plot, the Fermi surfaces for $\varepsilon_F = 0.1 \varepsilon_0$, $\varepsilon_F = \varepsilon_0$, and $\varepsilon_F = 10 \varepsilon_0$.

(2) Using the Sommerfeld expansion, compute the heat capacity for a two-dimensional electron gas, to lowest nontrivial order in the temperature T .

(3) ${}^3\text{He}$ atoms consist of an odd number of fermions (two electrons, two protons, and one neutron), and hence is itself a fermion. Consider a kilomole of ${}^3\text{He}$ atoms at standard temperature and pressure ($T = 293$, K, $p = 1$ atm).

- (a) What is the Fermi temperature of the gas?
- (b) Calculate $\mu/k_B T$ and $\exp(-\mu/k_B T)$.
- (c) Find the average occupancy $n(\varepsilon)$ of a single particle state with energy $\frac{3}{2}k_B T$.

(4) For ideal Fermi gases in $d = 1, 2$, and 3 dimensions, compute at $T = 0$ the average energy per particle E/N in terms of the Fermi energy ε_F .

(5) Obtain numerical estimates for the Fermi energy (in eV) and the Fermi temperature (in Kelvin) for the following systems:

- (a) conduction electrons in silver, lead, and aluminum
- (b) nucleons in a heavy nucleus, such as ${}^{200}\text{Hg}$

(6) Show that the chemical potential of a three-dimensional ideal nonrelativistic Fermi gas is given by

$$\mu(n, T) = \varepsilon_F \left[1 - \frac{\pi^2}{12} \left(\frac{k_B T}{\varepsilon_F} \right)^2 - \frac{\pi^4}{80} \left(\frac{k_B T}{\varepsilon_F} \right)^4 + \dots \right]$$

and the average energy per particle is

$$\frac{E}{N} = \frac{3}{5} \varepsilon_F \left[1 + \frac{5\pi^2}{12} \left(\frac{k_B T}{\varepsilon_F} \right)^2 - \frac{\pi^4}{16} \left(\frac{k_B T}{\varepsilon_F} \right)^4 + \dots \right] ,$$

where $\mu_0(n)$ is the Fermi energy at $T = 0$. Compute the heat capacity $C_V(T)$ to terms of order T^3 . How does the T^3 contribution to the electronic heat capacity compare with the contribution from phonons?