

**Formulas:**

Relativistic energy - momentum relation  $E = \sqrt{m^2 c^4 + p^2 c^2}$  ;  $c = 3 \times 10^8 \text{ m/s}$

Electron rest mass :  $m_e = 0.511 \text{ MeV}/c^2$ ; Proton :  $m_p = 938.26 \text{ MeV}/c^2$ ; Neutron :  $m_n = 939.55 \text{ MeV}/c^2$

Planck's law :  $u(\lambda) = n(\lambda) \bar{E}(\lambda)$  ;  $n(\lambda) = \frac{8\pi}{\lambda^4}$  ;  $\bar{E}(\lambda) = \frac{hc}{\lambda} \frac{1}{e^{hc/\lambda k_B T} - 1}$

Energy in a mode/oscillator :  $E_f = nhf$  ; probability  $P(E) \propto e^{-E/k_B T}$

Stefan's law :  $R = \sigma T^4$  ;  $\sigma = 5.67 \times 10^{-8} \text{ W/m}^2 \text{K}^4$  ;  $R = cU/4$  ,  $U = \int_0^\infty u(\lambda) d\lambda$

Wien's displacement law :  $\lambda_m T = \frac{hc}{4.96 k_B}$

Photons :  $E = pc$  ;  $E = hf$  ;  $p = h/\lambda$  ;  $f = c/\lambda$

Photoelectric effect :  $eV_0 = (\frac{1}{2} m v^2)_{\max} = hf - \phi$  ,  $\phi \equiv$  work function

Compton scattering :  $\lambda' - \lambda = \frac{h}{m_e c} (1 - \cos \theta)$

Rutherford scattering:  $b = \frac{k q_1 q_2}{m_\alpha v^2} \cot(\theta/2)$  ;  $\Delta N \propto \frac{1}{\sin^4(\theta/2)}$

Constants :  $hc = 12,400 \text{ eV}\text{\AA}$  ;  $\hbar c = 1,973 \text{ eV}\text{\AA}$  ;  $k_B = 1/11,600 \text{ eV/K}$  ;  $ke^2 = 14.4 \text{ eV}\text{\AA}$

Electrostatics :  $F = \frac{k q_1 q_2}{r^2}$  (force) ;  $U = q_0 V$  (potential energy) ;  $V = \frac{kq}{r}$  (potential)

Hydrogen spectrum:  $\frac{1}{\lambda} = R(\frac{1}{m^2} - \frac{1}{n^2})$  ;  $R = 1.097 \times 10^7 \text{ m}^{-1} = \frac{1}{911.3\text{\AA}}$

Bohr atom:  $E_n = -\frac{ke^2 Z}{2r_n} = -\frac{Z^2 E_0}{n^2}$  ;  $E_0 = \frac{ke^2}{2a_0} = \frac{mk^2 e^4}{2\hbar^2} = 13.6 \text{ eV}$  ;  $E_n = E_{kin} + E_{pot}$ ,  $E_{kin} = -E_{pot}/2 = -E_n$

$hf = E_i - E_f$  ;  $r_n = r_0 n^2$  ;  $r_0 = \frac{a_0}{Z}$  ;  $a_0 = \frac{\hbar^2}{mke^2} = 0.529\text{\AA}$  ;  $L = mvr = n\hbar$  angular momentum

X - ray spectra:  $f^{1/2} = A_n(Z - b)$  ; K :  $b = 1$ , L :  $b = 7.4$

de Broglie :  $\lambda = \frac{h}{p}$  ;  $f = \frac{E}{h}$  ;  $\omega = 2\pi f$  ;  $k = \frac{2\pi}{\lambda}$  ;  $E = \hbar\omega$  ;  $p = \hbar k$  ;  $E = \frac{p^2}{2m}$

group and phase velocity :  $v_g = \frac{d\omega}{dk}$  ;  $v_p = \frac{\omega}{k}$  ; Heisenberg :  $\Delta x \Delta p \sim \hbar$  ;  $\Delta t \Delta E \sim \hbar$

Wave function  $\Psi(x,t) = |\Psi(x,t)| e^{i\theta(x,t)}$  ;  $P(x,t) dx = |\Psi(x,t)|^2 dx =$  probability

Schrodinger equation :  $-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V(x)\Psi(x,t) = i\hbar \frac{\partial \Psi}{\partial t}$  ;  $\Psi(x,t) = \psi(x)e^{-i\frac{E}{\hbar}t}$

Time - independent Schrodinger equation :  $-\frac{\hbar^2}{2m} \frac{\partial^2 \psi}{\partial x^2} + V(x)\psi(x) = E\psi(x)$  ;  $\int_{-\infty}^{\infty} dx \psi^* \psi = 1$

$\infty$  square well:  $\psi_n(x) = \sqrt{\frac{2}{L}} \sin(\frac{n\pi x}{L})$  ;  $E_n = \frac{\pi^2 \hbar^2 n^2}{2mL^2}$

**Justify all your answers to all problems**

**Problem 1** (10 pts)

An electron is described by the wavepacket

$$\psi(x,t) = \int dk e^{i(kx - \omega(k)t)} a(k)$$

with  $a(k)=A$  for  $k$  in the interval  $(k_1, k_2)$  and  $a(k)=0$  outside that interval.  $k_1=99\text{\AA}^{-1}$ ,  $k_2=101\text{\AA}^{-1}$  and  $A$  is a constant.

- Find an expression for  $\psi(x, t = 0)$ . What is the most probable position for this electron at  $t=0$ ?
- Estimate the uncertainty in the position of this electron at  $t=0$ , in  $\text{\AA}$ .
- Estimate the most probable position for this electron at  $t=1\text{ s}$ .
- Estimate the group velocity of this electron, expressed as  $v/c$ .

**Problem 2** (10 pts)

An electron is confined to move in a one-dimensional potential of the form

$$V(x) = V_0 \frac{x^2}{x_0^2}$$

where  $V_0=5\text{ eV}$  and  $x_0=0.01\text{ \AA}$  and  $x$  is the position coordinate.

- Assume this electron is known for certain to be within a distance of  $0.01\text{ \AA}$  from the origin ( $x=0$ ). Give a rough estimate of (i) its potential energy and (ii) its kinetic energy, both in  $\text{eV}$ .
- Assume now the electron is in a state that minimizes its total energy (kinetic plus potential). Estimate the uncertainty in its position and its potential and kinetic energies, in  $\text{eV}$ .

Hint: Use Heisenberg's uncertainty principle. Use  $\hbar^2 / m_e = 7.62\text{ eV \AA}^2$

**Problem 3** (10 pts)

- An electron is in the ground state of an infinite square well of length  $L$ . How much more likely is it to be found at the center of the well ( $x=L/2$ ) than at  $x=L/4$ ?
- An electron in an infinite square well is in a state of energy  $1000\text{ eV}$ . Give a lower bound for the length of this well, in  $\text{\AA}$ . Can you also give an upper bound?

**Justify all your answers to all problems**