Formulas_Quiz 4

Formulas and constants

Mass of electron = 9.1. 10^{-31} kg Charge on electron = $1.6.10^{-19}$ C Planck's Constant h= $6.626. 10^{-34}$ J.s = $4.136. 10^{-15}$ eV.s $\hbar = h/2\pi = 1.055.10^{-34}$ J.s = $6.582.10^{-16}$ eV.s $1 \text{ eV} = 1.6. 10^{-19}$ J Coulomb's constant $k = 1/(4\pi\epsilon_0) = 8.99.10^9$ N.m² / kg² Velocity of light c = 3.10^8 m/s Bohr's quantization for Angular momentum $mvr = n\hbar$ Bohr radius $a_0 = 0.529. 10^{-10}$ m 1 Rydberg (Energy required to ionize hydrogen atom) = 13.6 eVRydberg Constant R = $1.097. 10^7$ m⁻¹ Energy of photon E = hf For photon λ f = c Force due to Electric field : $\mathbf{F} = q\mathbf{E}$

Drag Force on drop of radius α and velocity ν in medium of viscosity η (always opposite to direction of ν): $D = 6\pi\alpha\eta\nu = C\nu$